Atomic Resolution Characterization of Pt Based Bi-Metallic Nano-Catalysts Using Aberration Corrected STEM

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The global energy shortage requires the substitute of new energy strategy and various type of new materials are investigated for energy conversion purpose, including thermoelectric, ferroelectric, superconductors, etc.[1,2]. Among them fuel cell batteries are of particular interest due to their high capacity, good stability and reliability. The oxygen reduction reaction (ORR) in the fuel cells plays an important role in the energy generation, but the cost of the cathode/anode catalysts, which mainly contains Pt or Pd, limit its application due to their high production cost. In order to reduce the cost for the usage of precious materials, abundant materials with relative good catalytic properties, such as Co, Ni, can be synthesized with the Pt or Pd to achieve equal or better catalytic performance than the pure Pt nano-particle catalysts. In this work, different ratio of Pt-Co, Pt-Ni and Pt-Co-Cu nanoparticles were synthesized by wet chemical methods. By carefully controlling the experimental parameters, octahedral metal alloys or dodecahedral core shell nanoparticles can be synthesized. The catalytic properties were measured and it was found that some of these nanoparticles have comparable catalytic properties to the commercial 60% Pt/Vulcan products.

Scanning transmission electron microscopy (STEM) with energy dispersive x-ray spectroscopy (EDS) and electron energy loss spectroscopy (EELS) were applied to study the microstructure of the catalysts. The atomic resolution Z-contrast imaging (the image contrast is proportional to Z^2 , where Z is atomic number) was achieved by JEOL ARM 200F electron microscope with probe spherical aberration corrector. Figure 1 shows the Z-contrast image of the Pt₃Co dodecahedral nanoparticles. From the image contrast it can be seen that the structure of the particles seems to be core-shell, where the outer bright shell has a distinct boundary with the dark core. Figure 2 shows the Z contrast image of two nanoparticles with their corresponding EDS mapping result. The distribution of Pt and Co clearly indicates a core shell structure, while the elemental analysis shows the ratio of Pt:Co is approximately 3:1. As the crystal lattice mismatch between Co and Pt is quite large, it is difficult to synthesize the stable bimetallic nanoparticles with pure Co core covered by Pt shells due to the possible strong strain at the Pt-Co interface. Figure 3 shows an EDS linescan result across one core-shell particle, where the thickness of the shell is measured about 1nm. Different from the reported core-shell structures, where Pt intensity is normally maximized at the edge and decrease at the center, in these Pt₃Co particles the Pt signal reaches its maximum at the center and Co signal has a minimum at the same position. The possible explanation would be the composition of core is the Pt and Co alloy rather than pure Co cores, and the Co L edge electron energy loss spectra approve the alloy structure by comparing with Co metal and alloy reference spectra [3]. DFT calculation is also used to model the Pt shell structure and explains their good catalytic properties.

The different ratio of Pt:Co would generate bimetallic alloy with different shapes. Same synthesis method was applied to PtCo_{1.5} and the resulting nanoparticles were found to be octahedral and Pt was

found in the edge while Co in the center by EDS. More detailed analysis including EDS mapping, DFT calculation will also be discussed.

References:

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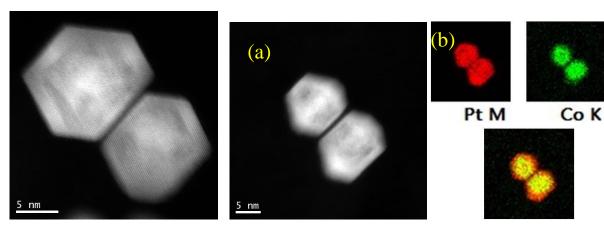
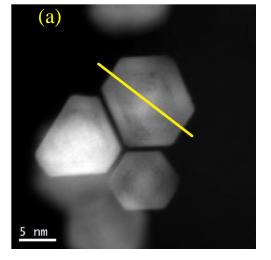


Figure 1. Z-contrast image of Pt₃Co dodecahedral nanoparticles

Figure 2. (a) Z-contrast image of Pt₃Co₂(b) EDS elemental maps of Pt, Co and a overlapped image



(b)

Figure 3. (a) Z-contrast image of Pt Co and the yellow line indicates the EDS linescan position, (b) EDS linescan intensity profiles of Co K peak and Pt M peak.