THE

BRITISH JOURNAL OF NUTRITION

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Papers intended for publication should be in doublespaced typing on one side of sheets of uniform size with adequate margins. Top copies only should be submitted, packed flat. The paper should be written in the English language, the spelling being that of the Oxford English Dictionary, and should, in general, be divided into the following parts: (a) Introductory paragraph, containing the reasons for publication of the work; (b) Experimental methods adopted: with chemical papers the experimental part will normally appear towards the end, but with other types of publication Methods should appear after Introduction; (c) Results: these should be given as concisely as possible, with the help of figures or tables; (d) Discussion: it is desirable that the presentation of the results and the discussion of their significance should be considered separately; (e) Summary: each paper must close with a summary in length not more than 5% of the previous text. This summary should aim at giving in the third person a complete picture in miniature of the entire article. The past tense should be used in referring to the author's experimental work. The present tense may be used where reference to existing knowledge is necessary, or where the author is stating what is shown or concluded. This change of tense should clearly differentiate the author's contribution from what is already known. The sequence in the summary should be the same as that in the paper. It is desirable to divide the summary into a series of numbered paragraphs or sentences giving, where relevant, the following information: a succinct account of the experimental work with essential facts concerning apparatus, chemicals, methods and animals; the results, singling out new information; the conclusions from the results. (f) References: these should be given in the text thus: Barnett & Robinson (1942), (Culbertson & Thomas, 1933); where a paper to be cited has more than two authors, the names of all the authors should be given when reference is first made, e.g. (Osborne, Mendel & Ferry, 1919); subsequent citations should appear thus: (Osborne et al. 1919). Where more than one paper by the same authors has appeared in one year the reference should be given as follows: Osborne & Mendel (1914a); Osborne & Mendel (1914b); or Osborne & Mendel (1914a, b); (Osborne & Mendel, 1914a, 1916; Barnett & Robinson, 1942).

References. At the end of the paper references should be given in alphabetical order according to the name of the first author of the publication quoted, and should include the authors' initials; the title of the paper should not be included. Titles of journals should be abbreviated in accordance with the system used in the World List of Scientific Periodicals (1934: 2nd ed. Oxford University Press). Examples of such abbreviations will be found in the current numbers of the British Journal of Nutrition and a useful list has recently been published in the Journal of Physiology (1945, 104, 232). References to books and monographs should include the town of publication and the name of the publisher, as well as the date of publication and the number of the edition to which reference is made. Thus:

Barnett, J. W. & Robinson, F. A. (1942). Biochem. J. 36,

Culbertson, C. C. & Thomas, B. H. (1934). Rep. Ia agric. Exp. Sta. 1933-4, p. 51.

Doisy, E. A., Somogyi, M. & Shaffer, P. A. (1923). J. biol. Chem. 55, xxxi.

Fairley, N. H. (1938). Nature, Lond., 142, 1156.

Hennessy, D. J. (1941). Industr. Engng Chem. (Anal. ed.), 13, 216.

King, H. (1941). J. chem. Soc. p. 338.

Osborne, T. B. & Mendel, L. B. (1914a). J. biol. Chem. 17, 325.

Osborne, T. B. & Mendel, L. B. (1914b). J. biol. Chem. 18, 1.
Osborne, T. B. & Mendel, L. B. (1916). Biochem. J. 10, 534.
Osborne, T. B., Mendel, L. B. & Ferry, E. L. (1919). J. biol. Chem. 37, 233.

Starling, E. H. (1915). Principles of Human Physiology, 2nd ed. London: Churchill.

Statistical Treatment of Data. In general the publication is not necessary of all the individual results of a number of replicated tests. A statement of the number of individual results, their mean value, and some appropriate measure of their variability, is usually sufficient.

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Chemical Formulas. These should be written as far as possible on a single horizontal line. With inorganic substances, formulas may be used, particularly in the experimental portion, at the discretion of the editors. With salts it must be stated whether or not the anhydrous material is used, e.g. anhydrous CuSO₄, or which of the different crystalline forms is indicated, e.g. CuSO₄.5H₂O, CuSO₄.H₂O.

Description of Solutions. Solutions of common acids, Cambridge Solutions of common acids, Cambridge Solutions are preferably defined in terms of normality (N) or molarity (M), e.g. N-HCl; 0·1 M-NaH₂PO₄. The term '%' must be used in its correct sense, i.e. g./100 g. of solution. 10% HCl means 10 g. of hydrogen chloride in 100 g. of aqueous solution, and should never be used to indicate a tenfold dilution of laboratory concentrated hydrochloric acid. For 'per cent by volume', i.e. ml./100 ml., the term '% (v/v)' may be employed. To indicate that g a given weight of substance is contained in 100 ml. of solution, the term '% (w/v)' (weight per volume) may be used.

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