

DIVISION XII / COMMISSION 14 / WORKING GROUP ATOMIC DATA

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1. Energy levels, wavelengths, line classifications, and line structure

The references cited in this section are mostly papers on original laboratory research; compilations and data bases are covered in another section. The references, ordered by atomic number and spectrum, are given in parentheses following the spectral notations. References including experimental data on line structure, hyperfine structure (HFS) or isotope structure (IS) are also included.

Li I [181], C I [136, 247], N I HFS,IS [109], Ne II IS [182, 125, 128], Ne III [182, 129], Ne IV [92], Mg I IS [212], Al VI HFS [45], S VII–XIV [146], K II [193], Sc I HFS [135], Cr I IS [87], Mn I HFS [39], Mn II HFS [40], Fe XV–XIX [161], Kr I [214], Sr I HFS,IS [53], Nb I HFS [133, 134], Cs I HFS [55], La I HFS [22, 89], La II HFS [220], Pr I HFS [88], Pr II HFS [90, 91], Nd II HFS,IS [122, 157, 198], Nd III [199], Nd IV [256, 257], Sm II HFS [156], Eu III [258], Gd I IS [107], Tm I HFS [21], Tm IV [165], Yb I HFS [56], Yb III [183], Hf II [155], Ta I HFS [106], Pb I IS [252].

The references for elements heavier than Ni ($Z > 28$) are limited to the first three or four spectra only, these data being of most interest for solar and stellar spectroscopy. The references of the lighter elements are also incomplete, the selection being limited to those of highest astrophysical interest. The data in a number of the references above include and/or supersede all or most of the previously data for the indicated spectrum.

Current analyses of high-resolution laboratory spectra (energy levels, wavelengths) is ongoing at Lund, Sweden (transition and rare-earth elements), London, UK (iron-group elements), NIST, USA (transition and rare-earth elements, HFS), Troitsk, Russia (heavy elements), and Meudon & Orsay, France (transition and rare-earth elements, theory).

2. Wavelength standards

Ritz wavelengths of forbidden lines of [Fe II], [Ti II], and [Cr II] have been measured using energy levels derived using Fourier transform spectroscopy (FTS) [10]. Accurate wavelengths of spectral lines in iron-group elements have been measured using FTS with uncertainties of around 10^{-5} nm [11]. The most accurate frequency standards are now being measured using laser spectroscopy with calibration from a laser frequency comb. Frequencies with uncertainties of less than 1 MHz have been measured for H I [99], Mg I [212], K I [73], Rb I [160], Sr I [53], Cs I [77, 94], and In II [250] using this technique.

Wavelengths of spectral lines from a Th/Ar hollow cathode lamp suitable for calibration of astronomical spectrographs have been published by various authors [117, 154, 172]. A correction to the wavelength scale of Ar I published by Whaling *et al.* in 2002 [253] is given in [213].

3. Transition probabilities

The references listed in section 7 are for the period 2005 - 2008. The transition-probability data in these references were obtained by both theoretical and experimental methods. The references for elements heavier than Ni ($Z > 28$) are limited to the first three or four spectra only. All papers contain a significant amount of numerical data, normally for more than ten spectral lines. Extensive results of atomic structure calculations are also given in reference [81].

4. Larger compilations, reviews, conference proceedings

The following compilations on wavelengths and energy levels have been published by the Atomic Spectroscopy group at NIST: Be II [124], B I [130], Ne II [128], Ne III [129], Ne VII [126], Ne VIII [127], K II-XIX [217], Ga I-XXXI [227], Kr I-XXXVI [211], Rb I-XXXVII [215], W I [131], W II [131], Hg I [210], Fr I [216]

In addition to these comprehensive compilations, the *Handbook of basic atomic spectroscopic data* [218] contains a selection of the strongest spectral lines of neutral and singly-ionized atoms of all elements from hydrogen to einsteinium. Compilations by other groups include He I [69, 169], B II [201], and a compilation of coronal lines [76].

The following additional major compilations of transition probability data have been published during the latest three year period: Na-like to Ar-like sequences [81], $^4\text{He I}$ [70], Fe I and Fe II [85], C I, C II, N I and N II [254], (Erratum: [255]). Additional data can be found in *NIST Atomic Transition Probabilities*, 86.

A number of papers on atomic spectroscopic data are included in proceedings of the *Ninth International Colloquium on Atomic Spectra and Oscillator Strengths for Astrophysical and Laboratory Plasmas*, held in Lund, Sweden, August 2007. Invited papers are published in the *Physica Scripta T Series* (Wahlgren, Wiese & Beiersdorfer 2008) and poster papers are published in the on-line open access *Journal of Physics Conference Series* (Wahlgren, Wiese & Beiersdorfer 2008).

Papers on astrophysical data needs are included in the proceedings from the *International Conferences on Atomic and Molecular Data* (ICAMDATA, Meudon, France, October 2006; Beijing, China, October 2008) and the *Fifteenth International Conference on Atomic Processes in Plasmas* (NIST, March 2007). These proceedings contain review papers as well as descriptions of atomic and molecular databases.

5. Atomic spectroscopic data on the internet

The following databases of atomic spectra are available at NIST. Most of these have received major updates since the last triennial report.

- *Atomic Spectra Database*
<physics.nist.gov/PhysRefData/ASD/index.html>
- *Handbook of Basic Atomic Spectroscopic Data*
<physics.nist.gov/PhysRefData/Handbook/index.html>
- *Energy Levels of Hydrogen and Deuterium*
<physics.nist.gov/PhysRefData/HDEL/index.html>
- *Ground Levels and Ionization Energies for the Neutral Atoms*
<physics.nist.gov/PhysRefData/IonEnergy/ionEnergy.html>
- *Spectral Data for the Chandra X-ray Observatory*
<physics.nist.gov/PhysRefData/Chandra/index.html>

- *NIST Atomic Spectra Bibliographic Databases*

<physics.nist.gov/PhysRefData/ASBib1/index.html>, consists of three databases of publications on atomic transition probabilities, atomic energy levels and spectra, and atomic spectral line broadening. All three are updated on a frequent basis.

Additional on-line databases including significant quantities of atomic data include:

- The *MCHF/MCDHF Collection on the Web* (C. Froese Fischer *et al.*), <www.vuse.vanderbilt.edu/~cff/mchf_collection/> contains results of multi-configuration Hartree-Fock (MCHF) or multi-configuration Dirac-Hartree-Fock (MCDHF) calculations for hydrogen and Li-like through Ar-like ions, mainly for $Z \leq 30$. Data for fine-structure transitions are included.

- The *TOPbase and Opacity Projects* include transition probability and oscillator strength data for astrophysically abundant ions for $Z \leq 26$). A database is available at <vizier.u-strasbg.fr/topbase/topbase.html>. Revised opacities for stellar astrophysics have been made available during the current reporting period.

- The *Database on Rare Earths at Mons university* (DREAM) database <w3.umh.ac.be/~astro/dream.shtml> continues to be a relevant source of data for wavelengths, energy levels, oscillator strengths and radiative lifetimes for neutral, singly and doubly-ionized rare earth elements. New data have not been added to this database during the past three years.

- *CHIANTI*, an atomic database for spectroscopic diagnostics of astrophysical plasmas <www.solar.nrl.navy.mil/chianti.html>, contains atomic data and programs for computing spectra from astrophysical plasmas, with the emphasis on highly-ionized atoms. During the current reporting period additions to the database (v.5) include new physical processes and atomic data.

- The *Vienna Atomic Line Database* (VALD) web site <ams.astro.univie.ac.at/~vald/> allows users to extract atomic line data from a compilation of numerous sources according to element or presence in stellar spectra.

- The *bibl database* is a comprehensive bibliographic database on atomic spectroscopy with a search engine on various atomic parameters is available at the Institute of Spectroscopy, Troitsk, <das101.isan.troitsk.ru/bibl.htm>.

6. List of References

The references are identified by a running number. This refers to the general reference list at the end of this report, where the literature is ordered alphabetically according to the first author. Each reference contains one or more code letters indicating the method applied by the authors, defined as follows:

Theoretical methods

Q	quant. mech. calc.	QF	quant. mech. calc. forbidden lines
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Experimental methods

EI	energy levels	WI	wavelength
HFS	hyperfine structure	IS	isotope structure
L	lifetime	M	miscellaneous
TE	emission transition probabilities		

Other methods

CP	data compilation	CM	comments
R	relative value only	F	forbidden line

7. References on lifetimes and transition probabilities

Ac I: 194	Co XI: 7, 242, 244	Hf I: 155
Ac II: 194	Co XIII: 242, 246	Hf II: 144, 155
	Co XVI: 9	
Ag II: 44	Co XVII: 266	In I: 205
		In II: 113
Al XIII: 8	Cr I: 230	
	Cr II: 180	Ir I: 262
Ar I: 57, 123, 271	Cr XII: 68	Ir II: 262
Ar II: 206	Cr XIII: 238	
Ar VII: 239		K V - KVII: 32
Ar X: 41	Cs I: 65	
Ar XI-Ar XVIII: 170		Kr II: 59, 159
Ar XI: 139	Cu II: 185	
Ar XIV: 143, 142		Li II: 110
Ar XVII: 1	F I: 276	
	F VI: 49, 219, 248, 274	Lu III: 29
Au I: 84	F VIII: 110	
Au II: 30, 84		Mg I: 12, 95, 116
	Fe I-Fe XVI: 96	Mg II - Mg XI: 116
B II: 49, 201, 219, 274	Fe I: 35, 85	Mg V: 28, 63
B IV: 110	Fe II: ?, 52, 85, 102	Mg VI: 140
	Fe III: 25, 64, 240	Mg VIII: 177
Ba I: 72, 222	Fe IV: 174	Mg IX: ?
Ba II: 207	Fe VII: 265	
	Fe VIII: 273	Mn I: 37
Be I: 95, 274	Fe IX: 6, 245, 273	Mn II: 190
Be III: 110	Fe X: 229	Mn XIII: 98
	Fe XI: 229	
Bi III: 195	Fe XII: 270, 246	Mo I: 190
	Fe XIII: 14, 115	
Br I: 191, 276	Fe XIV: 42, 68, 229	N I: 17, 18, 19, 47, 46,
Br II: 191	Fe XV: 9, 171	224, 234, 254
	Fe XVI: 4, 2	N II: 101, 173, 226, 243, 254
C I: 79, 254	Fe XVII-Fe XXIII: 141	N IV: 49, 219, 241, 248, 274
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	Fe XXII: 112	
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Ca III: 14, 237	Fe XXIV: 268	Nd IV: 257
Ca X: 97	Fe XXV: 111, 119	
Ca XIII: 137		Ne I: 60
Ca XIV: 138	Ga I: 202	Ne II: 43
Ca XIX: 119	Ga II: 16, 114, 164	Ne III: 129
		Ne VII: 101, 219, 248, 274
Cd I: 101	Gd II: 100	
Cd II: 167		Ni II: 108
	H I: 36, 103	Ni VI: 118
Cl I: 184, 228, 231, 276	He I: ?, 54, 69, 104, 110	Ni XI: 5
		Ni XIV: 246

Ni XVI: 68	Re I: 188	Th IV: 203
Ni XVII: 9	Re II: 190, 189	
Ni XVIII: 266		Ti I: 38
Ni XIX: 3	S I: 62, 272	Ti II: 23
Ni XXVI: 277	S II: 80	Ti IV: 120, 200
	S III: 14	Ti VI: 168
O I: 20, 48, 223	S X: 232	Ti XVIII: 275
O II: 178, 225, 235	S XII: 177	Ti XIX: 221
O III: 204	S XIII: 71, 263	Ti XX: 176
O IV: 233	S XV: 153, 119	
O V: 15, 219, 241, 248, 274		Tl I: 33
O V-O VIII: 93	Sc XIX: 251	
		V I: 259
Os I: 196	Si I: 35	V II: 259
Os II: 196	Si II: 236	V III: 105
	Si III: 74	V V: 66, 121
P II: 14, 50, 75	Si IX: 149	
	Si X: 149, 177	W I: 131
Pb II: 195, 209, 208	Si XI: 27, 149	W II: 131
	Si XIII: 186	
Pd I: 261		Xe I: 61, 67
	Sm II: 145, 197	Xe II: 58
Pm II: 83		
	Sn II: 205	Yb II: 179
Pr II: 148		Yb III: 183
Pr IV: 51	Sr I: 150, 264	
	Sr II: 147, 179, 207	Zn I: 78, 95
Ra I: 34, 72, 194, 222		Zn II: 163
Ra II: 194	Ta I: 82	
	Tb III: 260	Zr II: 151, 158
Rb I - XXXVII: 215		Zr III: 162
	Tc I: 187	
	Tc II: 190	

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