Atomic-Resolution EELS Study of Titanium Dopant Effects of Ca₃Co₄O₉ Thin Film

Xuan Hu, Patrick Phillips, Serdar Ogut, Robert Klie

Department of Physics, University of Illinois at Chicago, Chicago, IL 60607

Thermoelectric materials have attracted significant attention over the last few decades. As one of the outstanding thermoelectric oxide materials, the incommensurately layered Ca₃Co₄O₉ (CCO) exhibits a high in-plane Seebeck coefficient and a high thermoelectric figure of merit at high temperature. Many studies have reported that substitutional doping will increase the Seebeck coefficient [1], but understanding of dopant effects on local structure and electronic properties is still lacking. Using electron energy-loss spectroscopy (EELS) and first-principle calculations, the atomic and electronic structure of Ti dopants can be measured and the dopant effects on the thermoelectric properties can be analysed.

Ti-doped CCO thin films ($Ca_3Co_{3.8}Ti_{0.2}O_9$) were deposited by Pulsed Laser Deposition (PLD). All the images and EEL spectra were acquired by using an aberration-corrected JEOL ARM200CF with a 200 kV cold field emission gun and post-column Gatan Enfina EELS spectrometer. Figure 1a-c) shows the EELS image of Co, Ti, and Ca signals, respectively. The integrated signal intensity of Ti *L*-edge has been used to determine the position of Ti dopants. We find that the Ti dopants mainly replace Co atoms in the Ca_2CoO_3 subsystem. We have analysed the near edge fine-structure of the Ti *L*-edge and compared it to the shape of reference spectra for Ti^{4+} and Ti^{3+} [2]. We determine the valence state of Ti to be 4+. Figure 1c) shows the Co L_3/L_2 ratio as a function of concentration ratio of Ti/Ca. As the Ti concentration increases, the intensity ratio of $Co L_3/L_2$ remains mostly unchanged. Since the Co white lines ratio is directly linked to the Co valence state [3], we determine the Co valence state of CoO_2 subsystem to be (3.4 ± 0.2) and the Co valence in the Ca_2CoO_3 subsystem to be (2.8 ± 0.2) . The results are close to the values of the Co valence of pristine CCO bulk [4]. This demonstrates that the Ti doping does not influence the Co valence, especially the mixed-valence state of Co in CoO_2 layer, which means that hole concentration in the *p*-type CoO_2 layers remains unchanged and the effect of Ti doping on the Seebeck coefficient should be negligible.[5]

The experimental results will next be tested using first principles DFT modelling. Figure 2a) shows the unit cell of pristine 5/3 CCO with 66 atoms, which is used in our first-principle calculations. Based on this structure, the Ti dopants are studied by substituting Ti atom for one of the Ca atoms and Co atoms. All the 12 Ca sites and 16 Co sites have been considered. The calculated total energies show that Ti dopants are preferred on the Co sites of Ca₂CoO₃ subsystem compared to the Co sites of the CoO₂ layer, which is in agreement with our experimental observation. Figure 2b) shows the partial density of states (PDOS) of angular –momentum resolved *d*-orbitals of the Ti atom, substituting the Co atom of the Ca₂CoO₃ subsystem, where all the Ti 3d orbitals are empty. In addition, the PDOS of the Ti p-orbitals shows that the Ti 3p orbitals are completely occupied. We can conclude that the valence of Ti dopant is 4+. However, the effects of Ti dopants and their locations on local electronic structure is still being examined.

In this presentation, we will discuss out latest EELS result combined with first-principle calculations to analyse how Ti dopants influence the thermoelectric performance of CCO thin film [6].

References:

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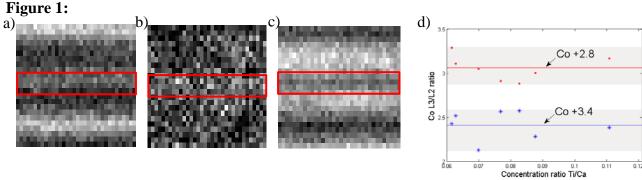


Figure 1. Ti doped $Ca_3Co_4O_9$ thin film along [110]: a-c) EELS images for Co, Ti, and Ca signals, respectively. The red rectangle is the area of CoO column of Ca_2CoO_3 subsystem. d) Intensity ratio of Co L_3/L_2 as a function of concentration ratio of Ti/Ca, for Co in CoO_2 layer (blue stars) and Co in CoO column of RS subsystem (red spots).

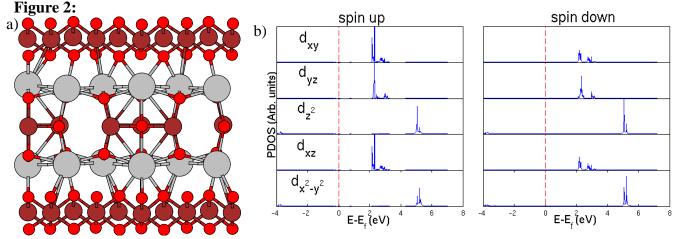


Figure 2. First-principle simulation of Ti doped Ca₃Co₄O_{9:} a) the unit cell of pristine 5/3 CCO with 66 atoms; b) the partial density of states (PDOS) of m-resolved d orbitals of Ti atom, which substitutes the Co atom of RS subsystem.