Transmission Electron Microscopy Characterization of IONPs Modified with Polyethylene Glycol Derivatives

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Iron oxide nanoparticles (IONPs) received great attention due to its potential use as platform for medical and pharmaceutical applications.[1,2] A protective layer on the IONPs surface is necessary for avoid the aggregation and, then, stabilize the dispersion. In some cases, this modification permits the interaction with others molecules or biomolecules. Wide coatings agents were used, some of them are polyethylene glycol derivatives. Polyethylene glycol (PEG) ensures the biocompatibility of these nanoparticles.[3]

The main goal of this work is to compare the size of IONPs covered with two different PEG derivatives: PEG-1kDa-(COCH₂CH₂COOH)₂ and PEG-3kDa-(CONHNH₂)₂). IONPs were synthetized by coprecipitation method. IONPs@PEG-1kDa-(COCH₂CH₂COOH)₂ and IONPs@PEG-3kDa-(CONHNH₂)₂) were obtained by *in situ* and *post-synthesis* methodologies, respectively. IONPs were characterized by attenuated total reflectance (ATR). In both cases, the vibration band of C-O-C around 1100 cm⁻¹ was observed. This fact suggests the presence of PEG chains on the surface of nanoparticles in comparison with uncoated IONPs.

A quasi-spherical morphology was detected through transmission electron microscopy (TEM). A better control of the size and shape results when the IONPs were obtained by *in situ* methodology. IONPs@PEG-1kDa-(COCH₂CH₂COOH)₂ has a core diameter of 6.7 ± 0.9 nm and 13.7 ± 1.3 nm for IONPs@PEG-3kDa-(CONHNH₂)₂. This study supports the influence of ligands addition during the process of nanoparticles formation.

Diffractograms permits to assign the different families of planes corresponding to spinel structure of iron oxides (magnetite, Fe₃O₄ and maghemite, γ-Fe₂O₃) comparing with IONPs with no organic layer obtained also by coprecipitation method. Families of planes (220) were observed in IONP@PEG-1kDa-(COCH₂CH₂COOH)₂ and planes (111) and (311) of the IONP@PEG-3kDa-(CONHNH₂)₂, by amplification of TEM images.

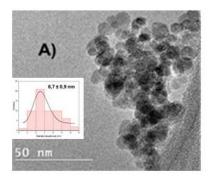
Conclusions

Synthesis of IONPs@PEG-1kDa-(COCH₂CH₂COOH)₂ obtained by *in situ* addition and IONPs@PEG-3kDa-(CONHNH₂)₂) by *post-synthesis* methodologies shows that the presence of polymeric ligand during the nanoparticles formation contributes significantly to produce smaller IONPs.

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References:

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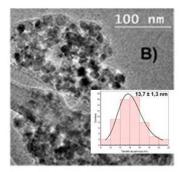


Figure 1. Transmission electron microscopy images and histograms of the IONP@PEG-1kDa-(COCH₂CH₂COOH)₂ obtained by *in situ* addition (**A**) and the IONP@PEG-3kDa-(CONHNH₂)₂ synthesized with *post-synthesis* addition (**B**).

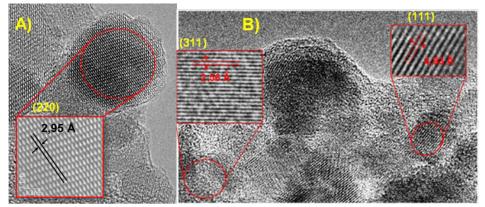


Figure 2. Approach of the TEM images, showing planes (220) in the IONP@PEG-1kDa-(COCH₂CH₂COOH)₂ (**A**) and planes (111) and (311) of the IONP@PEG-3kDa-(CONHNH₂)₂ (**B**).

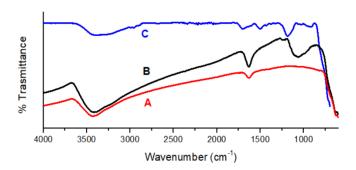


Figure 3. Attenuated Total Reflectance spectra of uncoated IONPs (**A**), IONP@PEG-3kDa-(CONHNH₂)₂ (**B**) and IONP@PEG-1kDa-(COCH₂CH₂COOH)₂ (**C**).

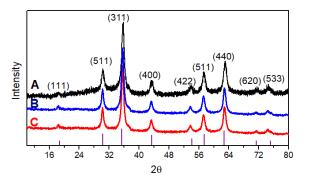


Figure 4. Diffractograms of the IONP@PEG-1kDa-(COCH₂CH₂COOH)₂ (**A**), the uncoated IONP (**B**) and the IONP@PEG-3kDa-(CONHNH₂)₂ (**C**).